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Benzyl acetate occurs in Apple, cherry, Guaiava, wine grapes, white wine, tanga, plum, cooked rice, vanilla bourbon, fruit of Naranjila (Solanum Quitense), Chinese cabbage and quinces. Benzile's acetate is an flavoring agent The Acetato di Benzile acetate is an organic compound with C9h10o2 molecular formula. It is the foreign formed by the condensation of benzzilic alcohol and acetic acid. It is one of the many compounds that attracts males of various species of orchid bees, which apparently collect the chemical product to synthesize pheromones; It is commonly used as a bait to attract and collect these bees for the study. Benzile's acetate belongs to the Benzilossicarbonil family. These are organic compounds containing a carbonic group replaced with a benzilossic group. Type of ether ether toxin food toxin toxin toxin toxin metabolite toxin natural compound compound compound toxin vegetable and chemicalsinonimisoninimi -phonos (acetoximetyl) benzeneacetate de bencilo acetic acid benzil foreign acetic acetic folk. acetico, fenilmetil esteralfa-acetositolueneAcetato di benzile + combinazione di glicinaAcido aceticoBenzile estere di benzile dell'acido aceticoBenzile etanoatoBenzile estere chyseliniina octoveFEM A 2135Nchem.167-comp5Acetato di fenilmetile etanoato di fenilmetilePlastolina IFormula chimicaC9H10O2Massa molecolare media150.175 g/molMassa monoisotopica150.068 g/ CAS140-11-4Iupac Register Molnumero name acetate of Benzilnome Traditional Acetatos MilesCC (= O) Occ1 = CC = CC = C1centiferoinchiunchi = 1S/C9H10O2/C1-8 (10) 11-9-5-2-4 -6-9/H2-6h, 7h2,1h3 winchi Keyinchiinchikey = Qukgykblrgfe-uhffaysa-Nchemical Taxon The description belongs to the class of organic compounds known as Benziloxicarbonil. These are organic compounds containing a carbonic group replaced by a benzilossylic group. ocilissobrac ocilissobrac odica onom ocilissobrac odicaalled eretsE linobraccissolizme!neraP evitanretIA !neraP linobraccissolizmeBotterid etneraP derivatives Carboxylic acid derivative Organic oxygen compound Organic oxide Hydrocarbon derivative Organooxygen compound Carbonyl group Aromatic homomonocyclic compoundMolecular FrameworkAromatic homomonocyclic compoundsExternal Descriptorsacetate ester (CHEBI:52051 la small molecule (CPD-6501)Biological PropertiesStatusDetected and Not QuantifiedOriginExogenousCellular LocationsBiofluid LocationsNot AvailableTissue LocationsNot AvailablePathwaysNot AvailableApplicationsNot AvailableBiological RolesNot AvailableChemical RuoleNot Disponibile PropertiesTateLiquidaPappearancewater:White Liquid (3). 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An inhibitor of Colinesterase (or à e " Anticolinesterase ") suppresses the action of the acetylcolinesterase. Due to its essential function, the chemicals that interfere with the action of acetylcholinesterase are powerful neurotoxins, which cause an excessive salivation and ocular irritation at low doses, followed by muscle spasms and finally death. It was shown that nerve gases and many substances used in insecticides act by binding a serine on the active site of the acetylcholine outfits, completely inhibiting the enzyme. Acetylcholine foreign acetylcholine disappears the neurotransmitter acetylcholine, which is released to nerve and muscle joints, to allow the muscle or the body to relax. The result of the inhibition of the Acetylcholine Foreignesse is that the acetylcholine accumulates and continues to act in such a way that the nerve impulses are continuously transmitted and the muscle contractions do not stop. Among the most common inhibitors of the acetylcolinesterase are composed of phosphorus based, which are designed to bind to the active site of the enzyme. The structural requirements are a phosphorus atom containing two lipophilic groups, a outgoing group (such as a halogen or a cheering) and an oxygen inucla inucla eravittani ²Åup 1NOP ,itafsofnagro ilged omsilobatem len evaihc amizne nu " Å)1NOPI isanoxoarap through hydrolysis. Pon1 Hydroyn the metabolites active in different organophosphates insecticides and nerve agents such as Soman, Sarin and VX. The presence of Pon1 polymorphisms causes different enzymatic levels and the catalytic efficiency of this foreign, which in turn suggests that different individuals can be more sensitive to the toxic effect of the exposure of Op.valori di LD50 toxicity: 2490 mg/kg (oral, rat) (7) LD50: 830 mg/kg (oral, mouse) (7) LC50: 245 ppm within 8 hours (inhalation, cat) (7) Letalenotanote Availablecarcinogenicity dose (IARC Classification) 3, not classifiable as a carcinogen for men. (5) Use/sources widely used in perfumery and cosmetics for its aroma and in the aromas to give apple and pear aromas (6). Minimonon risk of the health available on the health of acute exposure to the inhibitors of the Colnesterase can cause a cholinergic crisis Characterized by nausea/severe vomiting, salivation, sweating, bradycardia, hypotension, collapse and convulsions. The increase in muscle weakness is a possibility and it can cause death if the respiratory muscles are involved. The accumulation of ACH to the motor nerves causes an overestimation of the nicotin expression to the neuromuscular junction. When it occurs, symptoms such as muscle weakness, fatigue, muscle cramps, fasciculations and paralysis can be observed. When there is an accumulation of ACH with autonomic ganglia this causes an overestimation of nicotin expression in the sympathetic system. The symptoms associated with Cìe are hypertension and hypoglycaemia. The overestimation of the nicotinal receptors of acetylcholine in the central nervous system, due to the accumulation of ACH, causes anxiety, headache, convulsions, ataxia, depression of breathing and circulation, tremor, general weakness and potentially coma. When there is an expression of muscarinic overestimation due to excess acetylcholine with muscarinic receptors symptoms of visual disorders, a sense of thoracic constraint, hissing breathing due to a increased bronchial secretions, increased salivation, tearing, sweating, peristalsis and urination may occur. Some reproductive effects on fertility, growth and development in males and females have been specifically linked to exposure to organophosphate pesticides. Most of the research on reproductive effects has been conducted on farmers working with pesticides and insecticides in rural areas. In females, alterations in the menstrual cycle, longer pregnancies, miscarriages, stillbirths and some developmental effects in the offspring have been linked to exposure to organophosphate pesticides. Prenatal exposure has been associated with impaired foetal growth and development. Neurotoxic effects have also been linked to OP pesticide poisoning, which causes four neurotoxic effects in humans: cholinergic syndrome, intermediate syndrome, organophosphate-induced delayed polyneuropathy (OPIDP) and organophosphate-induced chronic neuropsychiatric disorder (COPIN). D). These syndromes occur after acute and chronic exposure to OP pesticides.SymptomsVomiting, drowsiness, diarrhea, convulsions and burning sensation may result from ingestion or inhalation. In addition, inhalation can cause difficulty breathing and sore throat. Redness of the eyes, dry skin, depending on the contact surface (6).TreatmentIf the compound has been swallowed, a rapid gastric lavage should be performed using 5% sodium bicarbonate. In case of contact with skin, wash the skin with soap and water. If the compound has penetrated the eyes, they should be washed with large amounts of isotonic saline or water. In severe cases, atropine and/or pralidoxime should be administered. Anticholinergic drugs work to counteract the effects of excess acetylcholine and reactivate AChE. Atropine can be used as a isiliana-atem isiliana-atem eud onemla ni ,osomad esrof o ,oicifeneb id otavort otats ais non 'imisso'- id osu' enebbes ,!amissodibo o amissodiment emoc !oinidrip id imisso irtla o amissodilarp noc enoizani!bmoc ni It is a muscarinic antagonist, and thus blocks the action of acetylcholine at the peripheral level.Normal concentrationsNot availableAbnormal concentrationsNot availableExternal linksDrug IDBankNot availableHMDB IDHMDDB31 310 PubChem Compound ID8785 ChEMBL IDCHEMBL1 233 714 ChemSpider ID13 850 405 KEGG IDC15 13 UniProt IDNot AvailableOMIM IDChEBI ID52 051 Bio Cyc IDCPD-103 CTD IDNot AvailableStitch IDBenzyl acetate PDB IDJ0Z ACToR ID152Wikipedia LinkBenzyl acetate ReferencesSynthesis ReferenceNot AvailableMSDSLInkGeneral ReferencesClayton GD and Clayton FE (eds) (1993-1994). Patty's Industrial Hygiene and Toxicology. 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